

Reg. No.

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**B.E. / B.TECH. DEGREE EXAMINATIONS, MAY 2023**

Second Semester

**PH22253 – Engineering Materials***(Common to AE, ME, MN)***(Regulation 2022)****TIME: 3 HOURS****MAX. MARKS: 100**

COURSE OUTCOMES	STATEMENT	RBT LEVEL
CO 1	Gain knowledge about various ferrous and non ferrous alloys and their phase diagrams and analyze the qualities of non ferrous alloys	4
CO 2	To understand various heat treatment techniques and analyze qualities of alloy steels.	4
CO 3	Evaluate the concepts of semiconductors and understand the behavior of electrons in the semiconductors	3
CO 4	Summarize basics of superconducting, magnetic and dielectric properties and few of the technologies and applications	3
CO 5	Enhance knowledge about various new engineering materials	3

**PART- A (20 x 2 = 40 Marks)**

(Answer all Questions)

	CO	RBT LEVEL
1. What is a solid solution? Give Examples.	1	2
2. Define Hume Rothery's Empirical rule for the substitutional solid solution.	1	3
3. Define Isomorphous system.	1	2
4. What is the maximum number of Phases that can coexist in equilibrium in a three component system?	1	3
5. Calculate atomic percent of carbon in mild steel containing 0.2 wt% of carbon.	2	3
6. Why is nitriding important?	2	2
7. State Fick's law of Diffusion.	2	2
8. Give the composition of low, medium and high carbon steel.	2	3
9. Graphically show the variation of conductivity of semiconductor with temperature.	3	3
10. Compare Direct and Indirect Band Gap Semiconductors.	3	3
11. In a solid, consider the energy level lying 0.05eV above Fermi level. What is the probability of this level being occupied by an electron at 300K?	3	3
12. What are the applications of Hall Effect?	3	2
13. Evaluate electronic Polarisability of Neon atom of radius 0.158nm.	4	3
14. Define Dielectric constant. Give its importance.	4	3
15. Mention the energies involved in origin of domains in ferromagnetic material.	4	2
16. Prove that superconductors are diamagnetic.	4	3
17. Give various types of ceramics.	5	2
18. What are the advantages of using metallic glasses as transformer core material?	5	3
19. Mention different forms of nano materials.	5	2
20. What are the types of Shape Memory Alloys?	5	2

**PART- B (5 x 10 = 50 Marks)**

	Marks	CO	RBT LEVEL
21. (a) Describe in detail the applications of Phase rule in study of Phase Diagram.	(10)	1	4
<b>(OR)</b>			
(b) What is Binary Phase diagram? Explain in detail about binary Isomorphous system and region present in it.	(10)	1	4
22. (a) Draw the Fe-C equilibrium diagram and label all the phases and their microstructures at Various temperatures for 0.8% C.	(10)	2	4
<b>(OR)</b>			
(b) Discuss about the various alloying elements in steel.	(10)	2	4
23. (a) Derive an expression for density of electrons in the conduction band of an intrinsic semiconductor.	(10)	3	3
<b>(OR)</b>			
(b) What is Hall effect? Discuss the theory of Hall effect for a N-type semiconductor and derive an Expression for Hall coefficient. How will you determine Hall Coefficient Experimentally?	(10)	3	3
24. (a) What is meant by Internal field and how it is calculated for a cubic Structure?	(10)	4	3
<b>(OR)</b>			
(b) Explain the Phenomenon of super conductivity and the properties exhibited by the super conductors.	(10)	4	3
25. (a) With a neat diagram describe the production, properties and applications of Metallic Glass.	(10)	5	2
<b>(OR)</b>			
(b) What are Nano materials? Explain Production, properties and applications of nanomaterials.	(10)	5	2

**PART- C (1 x 10 = 10 Marks)**

(Q.No.26 is compulsory)

	Marks	CO	RBT LEVEL
26. Deduce an expression for electronic polarisability and ionic polarisability in a dielectric material.	(10)	4	3